metal-organic papers

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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.004 Å R factor = 0.035 wR factor = 0.094 Data-to-parameter ratio = 18.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis[μ_2 -phenylbis(2-pyridyl)phosphine]- $\kappa^3 N, N': P; \kappa^3 P: N, N'$ -bis[chlorocopper(I)]

The title dinuclear complex, $[Cu_2Cl_2(C_{16}H_{13}N_2P)_2]$, possesses a crystallographically imposed inversion center. Two Cu atoms are connected by two bridging phenylbis(2-pyridyl)phosphine ligands. Each Cu atom adopts a distorted tetrahedral geometry, coordinated by one Cl atom [Cu-Cl =2.3107 (6) Å], two N atoms [Cu-N = 2.065 (2) and 2.066 (2) Å] from one phenylbis(2-pyridyl)phosphine ligand and one P atom [Cu-P = 2.1757 (7) Å] from another ligand.

Comment

Phenylbis(2-pyridyl)phosphine (PhPPy₂) is an analog of di-2pyridylamine that usually binds to metal atoms through two pyridyl groups. Such a feature is found in, for example, the mononuclear cobalt(II) dichloride adduct (Ehrlich *et al.*, 1984). However, in PhPPy₂, the P atom is also able to bind to transition metals with low oxidation states. This ligand can therefore coordinate to metals in different modes. In chlorobis(triphenylphosphine)ruthenium(II) hexafluorophosphate (Shutte *et al.*, 1997), PhPPy₂ binds to the metal atom intramolecularly, but intermolecularly in rhodium dichloride complexes (Galdecki *et al.*, 1999).



A direct reaction of $[Cu_2(C_{16}H_{13}N_2P)_2(CH_3CN)_2](CIO_4)_2$ [its crystal structure has recently been reported by Zhang *et al.* (2006)] with ammonium chloride led to the isolation of the title complex, (I), as yellow crystals, by substituting the labile acetonitrile molecules with Cl⁻. In the centrosymmetric dinuclear complex of (I), two Cu atoms are connected by two bridging PhPPy₂ ligands in a head-to-tail fashion. Each Cu atom is coordinated by one Cl and two N atoms from one PhPPy₂ ligand, and by one P atom from another PhPPy₂ ligand, in a distorted tetrahedral geometry (Table 1). The intramolecular Cu···Cu separation of 3.744 (1) Å in (I) indicates no interaction between copper centers.

Experimental

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Solid $[Cu_2(C_{16}H_{13}N_2P)_2(CH_3CN)_2](ClO_4)_2$ (Zhang *et al.*, 2006) (0.0088 g, 0.0094 mmol) was dissolved in acetonitrile (2 ml), to which

Received 2 November 2006 Accepted 20 November 2006 a methanolic solution (1 ml) of ammonium chloride (0.0010 g, 0.0188 mmol) was added. The product, (I), crystallized as yellow crystals overnight in 80% yield.

Z = 2

 $D_{\rm v} = 1.519 {\rm Mg m}^{-3}$

 $0.34 \times 0.14 \times 0.10 \text{ mm}$

9241 measured reflections

3548 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.0513P)^2$

+ 0.1517P] where $P = (F_o^2 + 2F_c^2)/3$

 $\Delta \rho_{\rm max} = 0.44 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.24 \text{ e } \text{\AA}^{-3}$

 $(\Delta/\sigma)_{\rm max} = 0.001$

2901 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 1.64 \text{ mm}^{-1}$

T = 293 (2) K

Block, yellow

 $R_{\rm int} = 0.025$

 $\theta_{\rm max} = 27.5^{\circ}$

Crystal data

 $[Cu_2Cl_2(C_{16}H_{13}N_2P)_2]$ $M_{\star} = 726.49$ Monoclinic, $P2_1/n$ a = 9.305 (1) Åb = 10.857 (1) Å c = 15.746 (2) Å $\beta = 93.033 \ (1)^{\circ}$ V = 1588.5 (3) Å³

Data collection

Bruker APEX area-detector diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.760, \ T_{\max} = 0.849$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ wR(F²) = 0.094 S = 1.043548 reflections 190 parameters H-atom parameters constrained

Table 1

Selected geometric par	ameters (Å, °).
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Cu1-N1	2.065 (2)	Cu1-P1 ⁱ	2.1757 (7)
Cu1-N2	2.066 (2)	Cu1-Cl1	2.3107 (6)
N1-Cu1-N2	94.18 (7)	N2-Cu1-P1 ¹	115.99 (6)
N1-Cu1-P1 ⁱ	116.63 (6)	N2-Cu1-Cl1	103.04 (6)
N1-Cu1-Cl1	103.84 (6)	P1 ⁱ -Cu1-Cl1	119.42 (3)
C	. 1 . 1 . 1		

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

H atoms were placed at idealized positions (C-H = 0.93 Å) and were included in the refinement in the riding-model approximation, with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2003); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine



Figure 1

The molecular structure of (I), showing the atom-labeling scheme and displacement ellipsoids drawn at the 50% probability level [symmetry code: (i) 1 - x, 1 - y, 1 - z].

structure: SHELXL97 (Sheldrick, 1997); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2006).

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